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NCERT TEXTBOOK QUESTIONS SOLVED. 3.1. How would you determine the standard electrode potential of the system Mg 2+1 Mg? Ans: A cell will be set up consisting of Mg/MgSO 4 (1 M) as one electrode and standard hydrogen electrode Pt, H, (1 atm)H + /(l M) as second electrode, measure the EMF of the cell and also note the direction of deflection in the voltmeter.

NCERT Solutions For Class 12 Chemistry Chapter 3 ...

Electrochemistry InText Solution. Question 1: How would you determine the standard electrode potential of the system Mg 2+ |Mg?. Answer: We know that it is not possible to measure the electrode potential of a half cell. We can only measure the difference between electrode potentials of two half cells, which gives the cell potential of the cell.

Class 12 Chemistry Electrochemistry In Text Questions Solution

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Henry’s law constant for CO 2 in water is 1.67 x 10 8 Pa at 298 K. Calculate the quantity of CO 2 in 500 mL of soda water when packed under 2.5 atm CO 2 pressure at 298 K. Ans.: 2.8 The vapour pressures of pure liquids A and B are 450 mm and 700 mm of Hg respectively at 350 K. Calculate the composition of the liquid mixture if total vapour pressure is 600 mm of Hg.

NCERT Solutions For Class 12 Chemistry Chapter 2 Solutions

Question 3.11:Suggest a list of metals that are extracted electrolytically. Question 3.12:Consider the reaction: Cr 2 O 7 2– + 14H + + 6e – → 2Cr 3+ + 8H 2 O What is the quantity of electricity in coulombs needed to reduce 1 mol of Cr 2 O 7 2–? Question 3.14:Suggest two materials other than hydrogen that can be used as fuels in fuel cells.

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Chemistry 12 Intext Solution Unit 2 - PvdA

Question: 1.18 - An element with molar mass 2.7 `xx` 10-2 kg mol-1 forms a cubic unit cell with edge length 405 pm. If its density is 2.7 `xx` 10 3 kg m-3, what is the nature of the cubic unit cell?. Answer: By knowing the number of atom in the cubic unit cell of given lattice, its nature can be determined. Given density (d) `=2.7xx10^3 text(kg)m^(-3)` Molar mass (M) `=2.7xx10^(-3)`kg

12 Chemistry Solid State NCERT In Text Solution part 2

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