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(a) Organic chemistry may be defined as the chemistry of

hydrocarbons and its derivatives. (b) Vital

Force Theory is a

theory made by the

Scientist Berzelius in

1809 which assumed

that organic

compounds are only

formed in living cells

and it is impossible to

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prepare them in
laboratories.

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0.12g of an organic compound containing Phosphorus gave 0.22g of $Mg_2 P_2 O_7$ by the usual analysis.

Calculate the percentage of Phosphorus in the compound. Answer:

Here the mass of

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organic compound
taken = 0.12g Mass of
 $Mg_2P_2O_7$ formed =
0.22g . Now 1 mole of
 $Mg_2P_2O_7$ = 222 g
of $Mg_2P_2O_7$ = 62g
of Phosphorus. i. e.
222g of $Mg_2P_2O_7$...

Organic Chemistry: Some Basic Principles and Techniques ...

MCQs on Organic
Chemistry for NEET
Organic Chemistry is
the branch of science

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dealing with the study of carbon-containing compounds. Carbon has a unique property of catenation due to which it forms C-C covalent bond leading to long carbon straight-chain and branched-chain compounds.

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The organic compound p-nitroacetanilide is prepared from

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acetanilide through nitration. When acetanilide is treated with nitrating mixture that is a mixture of nitric acid and sulphuric acid p-nitroacetanilide is formed. Along with p-nitroacetanilide, o-nitroacetanilide is also formed as a minor product.

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